### **APPLICATION NOTE**



# Determination of total nitrogen and TKN (Total Kjeldahl Nitrogen) in water and wastewater

Kjeldahl application based on **ISO 5663-1984 -** Water quality — Determination of Kjeldahl nitrogen, **AOAC 973.48 -** Nitrogen Total in Water, Kjeldahl Method, 1973.







#### Introduction

Nitrogen occurs in various forms in nature. In the nitrogen cycle, it is repeatedly transformed and then often enters water and wastewater via fertilisers, the excretions of humans and animals or through waste products from manufacturing industries. For the determination of nitrogen in water and wastewater, the most relevant substances are ammonium, organic nitrogen, nitrate and nitrite. Depending on the standard and the purpose of the determination, the following values are analysed:

Total nitrogen = TKN (Total Kjeldahl Nitrogen) + Nitrate + Nitrite

Total organic nitrogen = TKN (Total Kjeldahl Nitrogen) - ammonium

Total inorganic nitrogen = (nitrate + nitrite) + ammonium

TKN (Total Kjeldahl Nitrogen) = Total organic nitrogen + ammonium

#### C. Gerhardt instruments:

- KJELDATHERM KT20 (alternativ TURBOTHERM)
- VAPODEST 200 500C
- VACUSOG/TURBOSOG

### Additional equipment:

- Mixer for homogenisation
- Analytical balance
- Watter bath
- Fume cupboard

The composition and nitrogen compounds are particularly important for wastewater treatment. In the course of wastewater treatment in municipal wastewater treatment plants, organic nitrogen is converted into inorganic nitrogen. The TKN value (Total Kjeldahl Nitrogen), for example, is important for biological wastewater treatment. It has to be determined during the individual process steps for purifying the wastewater in order to monitor and possibly adjust the process. The ammonium content is also commonly important for waterworks and wastewater treatment plants, e.g. to determine the degree of contamination of the discharged water. At the end of the water treatment process, predominantly inorganic nitrogen is present. With the method described in the Application Note, based on the **Kjeldahl method**, the **TKN (Total Kjeldahl Nitrogen)** can be determined. With the other available applications "**Determination of ammonium in aqueous samples**" and "**Determination of nitrate in aqueous samples**", the individual values relevant for water and wastewater analyses can accordingly be determined.

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### The method

# Sample preparation and planning

The first important step in this method is to select a reasonable sample size. This should be based on the concentration of the chemicals used and the nitrogen content of the sample. For this purpose, we have prepared a table to simplify the routine.

Sample amount[ml]	Concentration range N [mg/L]	
500	<1	
250	1-5	
100	5-35	
50	20-70	
20	50-175	
10	100-350	
5	300-700	

→ **App note**: For the use of larger sample quantities, glass sizes up to 800 ml can be used by C. Gerhardt.

### **Digestion**

Now the selected volume of the sample is pipetted into an appropriate large Kjeldahl tube. Here, special attention should be paid to homogeneity. Furthermore, accurate addition is of great importance, as this has a great influence on the precision of the results. Then the catalyst tablets and the sulphuric acid are added. The digestion takes 90 minutes until the digestion has become clear.

→ **App note**: Depending on the water sample type, these tend to foam. In this case, a slower temperature programme can be selected or anti-foaming tablets can be used.

### Distillation + Titration

In the next step, the digested water samples are transferred to the VAPODEST distillation unit. Here, depending on the model, the addition of the water, the alkali and the titration is done automatically. Also here, the boric acid used and the corresponding titration acid should correspond to the nitrogen concentration range. In this way, reasonable titration volumes and excellent precision are achieved.

→ **App note**: In water analytics, low nitrogen concentrations are usually to be expected. Here, a low concentration of boric acid (between 0.1 - 1 %) and a likewise low concentration of titration acid (0.01 - 0.05 N) should be used. This improves the resolution and the standard deviation.



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### Exemplary results for waste water samples (table 1)

Sample type	Sample amount	Theor. content	Measured result	Repeatability / standard deviation
Zeolite process waste water	200 ml	20 ppm	20.03 ppm	0.1245
NMP in waste water	10 ml	0.8 % N	0.796	0.01

# Exemplary values German standard methods for water, waste water and sludge analysis (table 2)

Sample type	Sample amount	Theor. content	Repeatability standard deviation	Comparison standard deviation
Surface water	1-200 ml	9.84 ppm	0.57 ppm	1.93 ppm
Industrial water drainage	1-200 ml	206 ppm	3.30 ppm	7.60 ppm
Municipal water treatment plant drainage	1-200 ml	23.3 ppm	0.93 ppm	3.18 ppm

### Conclusion

A wide range of nitrogen compounds are relevant for water and wastewater analysis, both organic and inorganic. With the applications optimised for water analysis and the VAPODEST distillation systems, all relevant values can be determined automatically. In addition, special equipment from C. Gerhardt, such as the larger Kjeldahl flasks, make the analysis easier for the user.

For detailed information or other applications please contact:

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